

Activity 1

A coworker has been working on doping silicon crystals with boron atoms to control the conductivity of silicon. They have used doping levels ranging from 10^{16} cm^{-3} to 10^{19} cm^{-3} . Your coworker's manager suggests increasing the boron concentration to 10^{21} cm^{-3} , but your coworker has been concerned about the possibility of boron aggregation during the rapid thermal anneal (~1 second at 1000 °C) used for dopant activation.

Boron aggregation means that implanted boron atoms could diffuse into each other and form clusters. To help your coworker, you decide to calculate the diffusion lengths for boron in silicon and compare the diffusion lengths to the average distance between boron atoms, which is given by

$$d_{\text{ave}} = (n)^{-1/3},$$

where n is the dopant concentration.

You know that the diffusion coefficient of boron atoms in single crystal Si has been measured to be $1.5 \times 10^{-18} \text{ m}^2 \text{ s}^{-1}$ at 1000 °C and $1.1 \times 10^{-16} \text{ m}^2 \text{ s}^{-1}$ at 1200 °C.

Also, given that the solubility limit of boron is 0.2 at.%, you calculate an upper limit for the boron concentration in silicon and compare that to the boron diffusion length during annealing. Note, the density of silicon is 2.33 g cm^{-3} .

Should your coworker increase the boron concentration?